SCIENTIFIC

Acetic Acid 6.1 M - 17.4 M (Glacial) Safety Data Sheet (SDS)

SECTION 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Acetic Acid 6.1 M - 17.4 M (Glacial)

Flinn Scientific, Inc. P.O. Box 219, Batavia, IL 60510 (800) 452-1261 Chemtrec Emergency Phone Number: (800) 424-9338

SECTION 2 — HAZARDS IDENTIFICATION

Hazard class: Flammable liquids (Category 3). Flammable liquid and vapor (H226). Keep away from heat, sparks, open flames, and hot surfaces. No smoking (P210).

Hazard class: Acute toxicity, oral (Category 5). May be harmful if swallowed (H303).

Hazard class: Acute toxicity, dermal (Category 4). Harmful in contact with skin (H312).

Hazard class: Skin corrosion or irritation (Category 1). Causes severe skin burns and eye damage (H314).

Hazard class: Sensitization, skin (Category 1). May cause an allergic skin reaction (H317). Avoid breathing mist, vapors or spray (P261).

Hazard class: Acute toxicity, inhalation (Category 3). Toxic if inhaled (H331).

SECTION 3 - COMPOSITION, INFORMATION ON INGREDIENTS

Component Name		CAS Number	Formula	Formula Weight	Concentration
Acetic acid (/link/d94a5cc31d8d4db68d9e759f21b1a8f9.aspx)	Buy Now (/link/d94a5cc31d8d4db68d9e759f21b1a8f9.aspx)	64-19-7	СН3СООН	60.05	>90%
Synonyms: Vinegar acid; Ethanoic acid					

SECTION 4 - FIRST AID MEASURES

Immediately call a POISON CENTER or physician (P310). If inhaled: Remove victim to fresh air and keep at rest in a position comfortable for breathing (P304+P340). If in eyes: Rinse cautiously with water for several m inutes. Remove contact lenses if present and easy to do so. Continue rinsing (P305+P351+P338). If on skin or hair: Immediately remove all contaminated clothing. Rinse skin with water (P303+P361+P353). Wash contaminated clothing before reuse (P363). If skin irritation or rash occurs: Get medical advice or attention (P333+P313). If swallowed: Rinse mouth. Do NOT induce vomiting (P301+P330+P331).

SECTION 5 — FIRE FIGHTING MEASURES	
Class II combustible liquid.	NFPA Code
Flash point: 40 °C (CC) Flammable limits: Lower: 4.0% Upper: 19.9% Autoignition Temperature: 464 °C	H-3
When heated to decomposition, may emit toxic fumes.	F-2
In case of fire: Use a tri-class dry chemical fire extinguisher (P370+P378).	R-0

SECTION 6 — ACCIDENTAL RELEASE MEASURES

Remove all ignition sources and ventilate area. Contain the spill with sand or other inert absorbent material, neutralize with sodium bicarbonate or calcium hydroxide, and deposit in a sealed bag or container. See Sections 8 and 13 for further information.

SECTION 7 - HANDLING AND STORAGE

Flinn Suggested Chemical Storage Pattern: Organic #1. Store with acids, anhydrides and peracids. Store away from nitric acid Store in a dedicated acid cabinet and away from any source of water. If an acid cabinet is not available, store in a Flinn Saf-CubeTM.

Always add acid to water.

Keep container tightly closed (P233). Keep cool (P235). Use only in a hood or well-ventilated area (P271).

SECTION 8 - EXPOSURE CONTROLS, PERSONAL PROTECTION

Wear nitrile, neoprene or butyl rubber gloves, protective clothing, and eye protection (P280). Wash hands thoroughly after handling (P264). Contaminated work clothing should not be allowed out of the workplace (P272). Use only in a hood or well-ventilated area (P271).

Boiling point: 118 °C

Melting point: 16.6 °C

Specific gravity: 1.05

Freezes at slightly below room temperature (17 °C).

Exposure guidelines: PEL 10 ppm (OSHA) TLV 10 ppm Ceiling 15 ppm (ACGIH)

SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES

Clear, colorless liquid. Strong vinegar-like odor. Soluble: Miscible with water, alcohol, glycerol, and ether. Vapor pressure: 15.7 mm Hg at 25 °C Vapor density: 2.07 Viscosity: 1.056 mPa.s at 25 °C

SECTION 10 - STABILITY AND REACTIVITY

Avoid contact with strong oxidizers, especially chromic and nitric acids. Incompatible with metals (except aluminum), alcohols, amines, permanganates, peroxides, hydroxides, oxides, phosphates, and carbonates

pH <2>

Shelf life: Good, if stored properly. Store in a dedicated acid cabinet.

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Signal Word

DANGER

SECTION 11 – TOXICOLOGICAL INFORMATION

Acute effects: Corrosive to body tissues and mucous membranes. Chronic effects: Respiratory tract inflammation; darkens skin. Target organs: Eyes, skin, teeth, respiratory tract.

SECTION 12 – ECOLOGICAL INFORMATION

May be toxic to aquatic life. Microtox test: If released in water or soil, acetic acid will biodegrade readily.

SECTION 13 - DISPOSAL CONSIDERATIONS

Please review all federal, state and local regulations that may apply before proceeding. Flinn Suggested Disposal Method #24a is one option.

SECTION 14 - TRANSPORT INFORMATION

Shipping name: Acetic Acid, Glacial. Hazard class: 8, Corrosive, Flammable Liquid. UN number: UN2789

SECTION 15 - REGULATORY INFORMATION

TSCA-listed, EINECS-listed (200-580-7). RCRA code D001, D002.

SECTION 16 — OTHER INFORMATION

This Safety Data Sheet (SDS) is for guidance and is based upon information and tests believed to be reliable. Flinn Scientific, Inc. makes no guarantee of the accuracy or completeness of the data and shall not be liable for any damages relating thereto. The data is offered solely for your consideration, investigation, and verification. The data should not be confused with local, state, federal or insurance mandates, regulations, or requirements and CONSTITUTE NO WARRANTY. Any use of this data and information must be determined by the science instructor to be in accordance with applicable local, state or federal laws and regulations. The conditions or methods of handling, storage, use and disposal of the product(s) described are beyond the control of Flinn Scientific, Inc. and may be beyond our knowledge. FOR THIS AND OTHER REASONS, WE DO NOT ASSUME RESPONSIBILITY AND EXPRESSLY DISCLAIM LIABILITY FOR LOSS, DAMAGE OR EXPENSE ARISING OUT OF OR IN ANY WAY CONNECTED WITH THE HANDLING, STORAGE, USE OR DISPOSAL OF THIS PRODUCT(S).

N.A. = Not available, not all health aspects of this substance have been fully investigated.

N/A = Not applicable

Consult your copy of the Flinn Science Catalog/Reference Manual for additional information about laboratory chemicals.

Revision Date: December 30, 2015